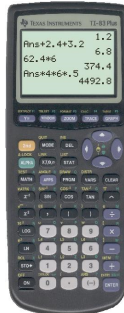


# 1.2 Scientific Notation and Significant Digits

## STANDARDS

**1.3** I can determine the precision of a measurement by identifying and utilizing significant digits.



## WARM UP!

Please do these problems in your notes to review.

- Give the SI base unit of measurement for each of the following quantities.
  - mass
  - length
  - time
  - temperature
- Convert the following numbers into scientific notation.
  - 85,000,000
  - 0.00019
- Put the following into decimal notation.
  - $8.72 \times 10^{-8}$
  - $3 \times 10^4$

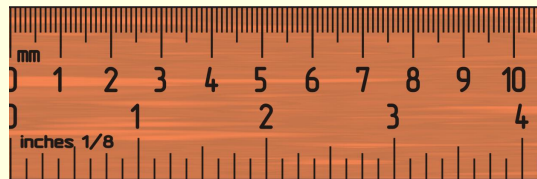
## WARM UP!

Please do these problems in your notes to review.

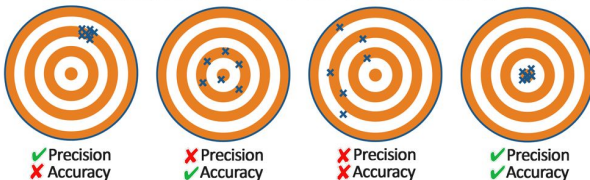
- Give the SI base unit of measurement for each of the following quantities.
  - mass **kg**
  - length **m**
  - time **s**
  - temperature **K**
- Convert the following numbers into scientific notation.
  - 85,000,000  **$8.5 \times 10^7$**
  - 0.00019  **$1.9 \times 10^{-4}$**
- Put the following into decimal notation.
  - $8.72 \times 10^{-8}$  **0.0000000872**
  - $3 \times 10^4$  **30,000**

## MEASUREMENT

A **measurement** is a comparison between an unknown quantity and a standard.



### PRECISION VS ACCURACY



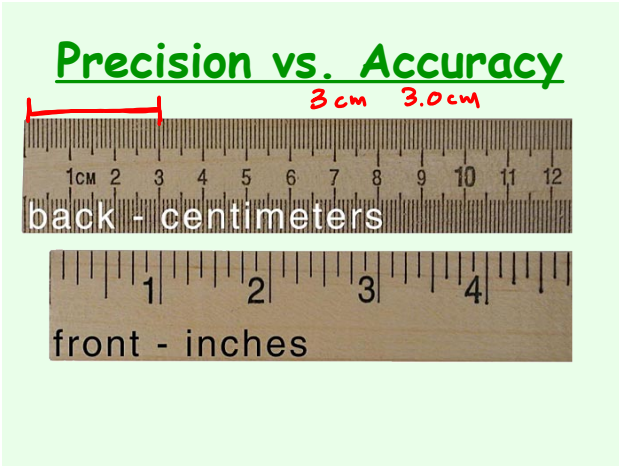
**Precision** refers to the closeness of two or more measurements to each other.

**Accuracy** refers to the closeness of a measured value to a standard or known value.

### Precision vs. Accuracy

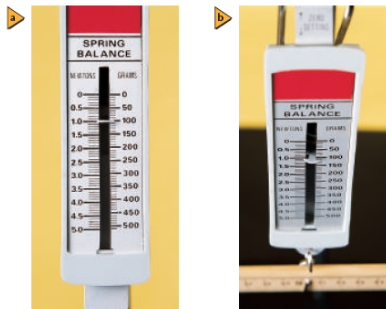


The precision of a measurement is 1/2 the smallest division of the instrument.



**Parallax** is the apparent shift in the position of an object when it is viewed from different angles.

**Parallax**



**WHY SIGNIFICANT DIGITS?**  
 "A measurement without any knowledge of its uncertainty is meaningless!"  
 - Walter Lewin, MIT

**SIGNIFICANT DIGITS**

Two athletes are trying to decide who has a faster time in the 100 m dash. One runner states that his time was 11 s and the other runner's time was 11.0 s. What is the difference between their times?

10.6 - 11.4

10.96 - 11.04

SIGNIFICANT DIGITS

How many significant digits are there in each number?

- 1) 0.0860 m
- 2) 172,000 g
- 3)  $1.7200 \times 10^5$  g

SIGNIFICANT DIGITS

All non-zero numbers are significant?

SIGNIFICANT DIGITS

Are all zeros significant?

- Leading? NEVER
- Tweeners? ALWAYS
- Trailing? SOMETIMES  
AFTER DEC. → YES  
BEFORE DEC → NO

SIGNIFICANT DIGITS

How many significant digits?

- a.) 0.0860 m (3)
- b.) 172,000 g (3)
- c.)  $1.7200 \times 10^5$  g (5)

8.34281 s

SIGNIFICANT DIGITSAdding / Subtracting

The number of decimal places after addition or subtraction is equal to the smallest number of decimal places in any of the individual terms.

SIGNIFICANT DIGITS

Solve the following problem using significant digits.

$$\textcircled{2} \quad \textcircled{1} \\ 3.86 \text{ m} + 2.4 \text{ m} = 6.26 \text{ m}$$

$\textcircled{6.3 \text{ m}}$

**SIGNIFICANT DIGITS****Multiplying / Dividing**

The number of significant figures after multiplication or division is equal to the number of significant figures in the least accurately known quantity.

**SIGNIFICANT DIGITS**

Solve the following problem using significant digits.

$$\overset{(4)}{409.2} \text{ km} / \overset{(3)}{11.4} \text{ s} = 35.89473684$$

$$35.9 \frac{\text{km}}{\text{s}}$$

$$35.89 \frac{\text{km}}{\text{s}}$$

# HOMEWORK

## Unit 1 Problems (6-7)